

Dissociation Constants for Bases at 25°C

Name	Formula	K_b
Ammonia	NH_3	1.80×10^{-5}
Aniline	$\text{C}_6\text{H}_5\text{NH}_2$	4.30×10^{-10}
Dimethylamine	$(\text{CH}_3)_2\text{NH}$	5.40×10^{-4}
Ethylamine	$\text{C}_2\text{H}_5\text{NH}_2$	6.40×10^{-4}
Hydrazine	H_2NNH_2	1.30×10^{-6}
Hydroxylamine	HONH_2	1.10×10^{-8}
Methylamine	CH_3NH_2	4.40×10^{-4}
Pyridine	$\text{C}_5\text{H}_5\text{N}$	1.70×10^{-9}
Trimethylamine	$(\text{CH}_3)_3\text{N}$	6.40×10^{-5}