Dissociation Constants for Bases at 25°C		
Name	**Formula**	**K_b**
Ammonia	NH_3	1.80×10^{-5}
Aniline	$\mathbf{C_6H_5NH_2}$	$\mathbf{4.30 \times 10^{-10}}$
n-Butylamine	$C_4H_9NH_2$	4.0×10^{-4}
***sec*-Butylamine**	$\mathbf{(CH_3)_2CHCH_2NH_2}$	$\mathbf{3.6 \times 10^{-4}}$
tert-Butylamine	$(CH_3)_3CNH_2$	4.8×10^{-4}
Codeine	$\mathbf{C_{18}H_{21}O_3N}$	$\mathbf{8.91 \times 10^{-7}}$
Diethylamine	$(CH_3CH_2)_2NH$	3.09×10^{-5}
Dimethylamine	$\mathbf{(CH_3)_2NH}$	$\mathbf{5.40 \times 10^{-4}}$
Ethylamine	$C_2H_5NH_2$	6.40×10^{-4}
Hydrazine	$\mathbf{H_2NNH_2}$	$\mathbf{1.30 \times 10^{-6}}$
Hydroxylamine	$HONH_2$	1.10×10^{-8}
Methylamine	$\mathbf{CH_3NH_2}$	$\mathbf{4.40 \times 10^{-4}}$
Morphine	$C_{17}H_{19}O_3N$	7.41×10^{-7}
Piperidine	$\mathbf{C_5H_{11}N}$	$\mathbf{1.32 \times 10^{-3}}$
Propylamine	$C_3H_7NH_2$	3.5×10^{-4}
Pyridine	$\mathbf{C_5H_5N}$	$\mathbf{1.70 \times 10^{-9}}$
Quinoline	C_9H_7N	6.31×10^{-10}
Triethanolamine	$\mathbf{C_6H_{15}O_3N}$	$\mathbf{5.75 \times 10^{-7}}$
Triethylamine	$(CH_3CH_2)_3N$	5.25×10^{-4}
Trimethylamine	$\mathbf{(CH_3)_3N}$	$\mathbf{6.40 \times 10^{-5}}$